

NEW Nelson Chemistry Alberta 20-30 Periodic Table (50 pack)

New and updated periodic table classroom packs to support *Nelson Chemistry Alberta 20-30*.

Nelson Chemistry Alberta 20-30 Periodic Table of the Elements

Table of Common Polyatomic Ions

Formula	Name
NO ₂ ⁻	nitrite
NO ₃ ⁻	nitrate
CO ₃ ²⁻	carbonate
SO ₄ ²⁻	sulfate
PO ₄ ³⁻	phosphate
ClO ₄ ⁻	perchlorate
ClO ₃ ⁻	chlorate
ClO ₂ ⁻	chlorite
ClO ⁻	hypochlorite
SO ₃ ²⁻	sulfite
SO ₄ ²⁻	sulfate
SO ₃	sulfur trioxide
SO ₂	sulfur dioxide
CO ₂	carbon dioxide
CO ₃ ²⁻	carbonate
CO	carbon monoxide
PO ₄ ³⁻	phosphate
PO ₃ ³⁻	phosphite
PO ₃	phosphorus trioxide
PO ₂	phosphorus dioxide
PO	phosphorus monoxide
AsO ₄ ³⁻	arsenate
AsO ₃ ³⁻	arsenite
AsO ₃	arsenic trioxide
AsO ₂	arsenic dioxide
AsO	arsenic monoxide
SbO ₄ ³⁻	antimonate
SbO ₃ ³⁻	antimonite
SbO ₃	antimony trioxide
SbO ₂	antimony dioxide
SbO	antimony monoxide
BiO ₄ ³⁻	bismutate
BiO ₃ ³⁻	bismutite
BiO ₃	bismuth trioxide
BiO ₂	bismuth dioxide
BiO	bismuth monoxide
VO ₄ ³⁻	vanadate
VO ₃ ³⁻	vanadite
VO ₃	vanadium trioxide
VO ₂	vanadium dioxide
VO	vanadium monoxide
CrO ₄ ²⁻	chromate
CrO ₃	chromium trioxide
CrO ₂	chromium dioxide
CrO	chromium monoxide
MnO ₄ ⁻	manganate
MnO ₃ ⁻	manganite
MnO ₃	manganese trioxide
MnO ₂	manganese dioxide
MnO	manganese monoxide
FeO ₄ ²⁻	ferrate
FeO ₃	iron(III) oxide
FeO ₂	iron(II,III) oxide
FeO	iron(II) oxide
CoO ₄ ²⁻	cobaltate
CoO ₃	cobalt(III) oxide
CoO ₂	cobalt(II,III) oxide
CoO	cobalt(II) oxide
NiO ₄ ²⁻	nickelate
NiO ₃	nickel(III) oxide
NiO ₂	nickel(II,III) oxide
NiO	nickel(II) oxide
CuO ₄ ²⁻	cuprate
CuO ₃	copper(III) oxide
CuO ₂	copper(II,III) oxide
CuO	copper(II) oxide
ZnO ₄ ²⁻	zincate
ZnO ₃	zinc(III) oxide
ZnO ₂	zinc(II,III) oxide
ZnO	zinc(II) oxide
AgO ₄ ²⁻	argentate
AgO ₃	silver(III) oxide
AgO ₂	silver(II,III) oxide
AgO	silver(I,II) oxide
HgO ₄ ²⁻	mercurate
HgO ₃	mercury(III) oxide
HgO ₂	mercury(II,III) oxide
HgO	mercury(II) oxide
PbO ₄ ²⁻	plumbate
PbO ₃	lead(IV) oxide
PbO ₂	lead(II,IV) oxide
PbO	lead(II) oxide
SnO ₄ ²⁻	stannate
SnO ₃	tin(IV) oxide
SnO ₂	tin(II,IV) oxide
SnO	tin(II) oxide
BiO ₄ ²⁻	bismutate
BiO ₃	bismuth(III) oxide
BiO ₂	bismuth(III,IV) oxide
BiO	bismuth(III) oxide
TeO ₄ ²⁻	tellurate
TeO ₃	tellur(IV) oxide
TeO ₂	tellur(IV) oxide
TeO	tellur(IV) oxide
PoO ₄ ²⁻	polonate
PoO ₃	polonium(IV) oxide
PoO ₂	polonium(IV) oxide
PoO	polonium(IV) oxide
AtO ₄ ²⁻	astatate
AtO ₃	astatine(IV) oxide
AtO ₂	astatine(IV) oxide
AtO	astatine(IV) oxide
FrO ₄ ²⁻	franciate
FrO ₃	francium(IV) oxide
FrO ₂	francium(IV) oxide
FrO	francium(IV) oxide

Nelson Chemistry Alberta 20-30 Data Tables

TABLES OF COMMON AQUEOUS IONS

FLAME COLORS OF ELEMENTS

TABLES OF SPECIFIC HEAT CAPACITIES

TABLES OF SOLUBILITY OF SOME COMMON IONIC COMPOUNDS IN WATER AT 298.15 K

ACID-BASE INDICATORS AT 298.15 K

DEFINED (EXACT) QUANTITIES

MEASURED (UNCERTAIN) QUANTITIES

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- Values from *Alberta Education Chemistry Data Booklet, Revised 2008* and *CRC Handbook for Chemistry and Physics*

Referenced Data Table information includes:

- Colours of common aqueous ions
- Some typical commercial reagent solutions
- Specific heat capacities of pure substances
- Flame colour of elements
- Selected SI prefixes
- Chemistry notation
- Solubility of some common Ionic compounds in Water at 298.15 K
- Acid-base indicators at 298.15 K
- Defined (exact) quantities
- Measured (uncertain) quantities

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